International Journal of Basic and Applied Chemical Sciences ISSN: 2277-2073 (Online) An Open Access, Online International Journal Available at http://www.cibtech.org/jcs.htm 2016 Vol. 6(4) October-December, pp.8-13/Madu et al.

Research Article

INVESTIGATION OF THE POSSIBLE IMPACT OF SEISMIC EXPLOSIVE ENERGY SOURCES ON THE TURBIDITY OF GROUNDWATER IN SAGBAMA, NIGER DELTA, NIGERIA

A.J.C. Madu¹, C.L. Eze² and *I.E Otuokere³

¹Department of Geology, Michael Okpara University of Agriculture, Umudike, Nigeria ²Institute of Geosciences and Space Technology, Rivers State University of Science and Technology, Port Harcourt, Nigeria

³Department of Chemistry, Michael Okpara University of Agriculture, Umudike, Nigeria *Author for Correspondence

ABSTRACT

Explosives contamination of groundwater has been a major environmental challenge. The possible impact of seismic explosive energy sources on the turbidity of groundwater in Sagbama, Niger Delta, Nigeria was investigated using 2,4,6-trinitrololuene (TNT) energy source. The energy source was the high explosive dynamite 2,4,6-trinitrololuene (TNT) and 6m Electric Detonators loaded in 5 hole pattern source array. The eleven upholes used in this study were drilled to 60m depth using the rotary method and flushed continuously for 20 minutes to enhance stability. The turbidity values were determined using a turbidimeter. A control sample was taken from the borehole stations by sampling a day before detonation of dynamite. Subsequently, sampling was carried out a day after dynamite detonation and then, on a fourth-nightly basis. The average turbidity value of the control (sample water before detonation) was 3.76 NTU. After dynamite detonation, the average measured turbidity values ranged from 2.91 to 3.80 NTU. The highest recorded average value of 3.80 NTU was obtained on day-57 after dynamite detonation. Thereafter the average turbidity value remained fluctuated. The lowest average value of 2.91 NTU was recorded on day-99. These variations from the control are not significant enough for dynamite to be said to have impacted on the turbidity value of the ground water. The control and test results values are both within the Federal Ministry of Environment, Housing and Urban Development (FMEnv&UD) limit of 10 NTU. Similar daily variations have been observed in areas where there was no dynamite detonation.

Keywords: Explosives, Contamination, 2,4,6-Trinitrololuene, Turbidity

INTRODUCTION

Explosion can be defined as the sudden release of energy. This energy may be obtained from nuclear or chemical reactions. Most explosives are composed of nitrogen compounds. The compounds oxidize to form small molecules such as H_2O , N_2 and CO_2 (Dimitrios *et al.*, 2011). Based on susceptibility to initiation, explosives can be classified as primary and secondary (Dimitrios *et al.*, 2011). Primary explosives are highly susceptible to explode and they can be used to initiate secondary explosives (Dimitrios *et al.*, 2011).

Explosives contamination of groundwater has been a major environmental challenge. The transportation of 2,4-dinitrotoluene (2,4-DNT) and 2,6-dinitrotoluene (2,6-DNT) have been evaluated (Dontsova *et al.*, 2009). Explosive contamination of ground water and soil were reported in Belgium after World War 1 ammunition destruction (Bausinger and Preuss, 2005).

The groundwater was contaminated with lead, arsenic, copper and nitroaromatics (Bausinger and Preuss, 2005). The production and detonation of explosives at the Werk Tanne ammunition site in Claudthal Zellerfeld, Germany caused a major pollution in that area (Eisentraeger *et al.*, 2007). It was reported that the site was polluted with polycyclic aromatic hydrocarbons (PAHs) and heavy metals (Eisentraeger *et al.*, 2007).

Agricultural and rangelands around Partex plant in the U.S was reported contaminated with Royal Demolition eXplosive (RDX), Trinitrotoluene (TNT) and (1,3,5,7-tetranitro-1,3,5,7-tetrazocane) HMX. This contamination was concentrated 10 meters just below the soil surface (Gray and McGrath, 2007). In

Research Article

Cambodia, Southeast Asia, contamination due to mortar bombs, rocket-propelled, artillery shells, cluster bombs, rifle grenades and aircraft bombs were reported (Bendinelli, 2009). Explosives may enter the environment during the process of production, storage, usage or disposal, resulting in the contamination of groundwater. Organisms in the environment will be influenced by the explosive compounds (Pennington and Brannon, 2002; Juhasz and Naidu, 2007). The rate and extent of transport and transformation of the explosive compounds are governed by the solubility, vapour pressure, Henry's law constant, soil properties, pH and weather conditions.

Explosive compounds have the tendency to generate toxic effects when the exposure takes a longer time. The effect of TNT explosion on mice, rats and dogs showed that it causes adverse health effect, including genotoxic and carcinogenic effects (ATSDR, 1995).

In the light of groundwater contamination by explosive, we hereby report the possible effect of seismic energy sources on the turbidity of groundwater in Sagbama, Niger Delta, Nigeria.

MATERIALS AND METHODS

Study Area: This study was carried out in Oil Mining Lease (OML) X in Niger Delta, Nigeria. The vegetation of the area is generally dominated by dense rain forest. The total area covered by this is study about 771.26 km².

Uphole Drilling: The uphole was drilled to 60m depth using the rotary method and flushed continuously for 20 minutes to enhance stability. Each hole was cased with perforated 6 inch PVC pipe. The uphole lithology was sampled every 5m or at the change in lithology. The grain size analysis was accomplished using the sieve method for the sands.

The result of the sieve analysis was used to calculate particle size, sorting, roundness and sphericity with the aim of estimating the porosity and permeability of the soil. No size analysis was carried out for the silt and clay. The source and receiver line number were used for identification of the uphole locations while the coordinates was verified using a handheld Global Positioning System (GPS) set. The Meridian Platinum GPS was used after calibrations. The elevations of the uphole points were determined using the same instrument (Table 1).

Table 1: Sagbama (SG) Monitoring Boreholes Showing Location x-y Coordinates and Elevation

Uphole No.	Uphole Location	Easting	Northing	Elevation (Z)		
SG1	2030/4070	397450.42	133107.70	18.70		
SG2	270/4070	397850.88	129103.76	19.80		
SG3	2110/4070	397450.72	125101.30	16.20		
SG4	1950/4070	397451.30	121100.00	28.10		
SG5	1790/4070	397450.50	117100.30	32.80		
SG6	2430/4230	401451.50	133101.60	18.14		
SG7	2270/4230	401450.82	129100.90	18.50		
SG8	2110/4230	401452.74	125102.84	26.70		
SG9	1950/4230	401450.50	121600.80	27.20		
SG10	1790/4230	401448.80	117100.50	31.80		
SG11	2430/4390	405450.82	133101.18	39.60		

Explosive Detonation: The energy source was the high explosive dynamite (TNT) and 6m Electric Detonators loaded in 5 hole pattern source array. In dry areas, the holes were thumped to a depth of 4m while in flooded/marshy terrain they were flushed to a depth of 6 meters. Each pattern hole was loaded with 0.4kg/l seismic, electric detonators (1 shot point = 5x0.4kg/l caps for five hole pattern). A total

International Journal of Basic and Applied Chemical Sciences ISSN: 2277-2073 (Online) An Open Access, Online International Journal Available at http://www.cibtech.org/jcs.htm 2016 Vol. 6(4) October-December, pp.8-13/Madu et al.

Research Article

amount of 84207 kg dynamite was detonated in 41, 949 source point in an area of 549.73 square kilometers of Sagbama area.

Water Sampling and Analysis: The groundwater was sampled from 11 upholes in Sagbama. The sample locations were geo-referenced using the survey coordinates of eastings and northings. The elevation of each sample location was determined by measurement of the Z-component of the coordinates.

The water samples were taken from the boreholes at a depth of 12m to 15m (Table I). Water samples were collected and analyzed from each uphole before the commencement of dynamite detonation. The water samples were collected using small bottles using with a rope.

The results of the nitrate analysis before the commencement of dynamite detonation served as control. The water was subsequently sampled fortnightly, that is 14 days for an interval of 2 months after the seismic acquisition process has been completed. The results obtained from the analysis of the samples during and after acquisition were compared with the results obtained before detonation and also with standard values.

Measurement of Turbidity: The turbidity was determined the same day the samples were taken. The power supply of the turbidimeter was switch on. 1 NTU button was pressed and was coincided with zero using the focusing template. 1 NTU button was pressed again and was coincided using the focusing template. A standard formazine solution of was placed on the turbidimeter in path of rays and the scale was brought to 9 NTU. The water sample was taken in a test and placed in a turbidimeter. The readings were recorded.

Measurement of Permeability: Coefficient of permeability was measured using the constant head permeameter. Soil sample was placed in the cylinder. A measurement was taken between the two tapings in the cylinder connected to the manometers. Water from the reservoir was allowed to flow through the sample at a constant rate.

As soon as the water begins to flow, the stop clock reading begins. Water flowing through the sample was collected with the measuring cylinder. The differences of heads from the manometers were measured. The time was recorded using the stop clock.

This was repeated for two other samples. Coefficient of permeability (K) was calculated using Darcy's Law (equation 1)

Where Q = Quantity of water passing through sample during time $t(cm^3/sec)$,

A = Cross - sectional area of sample (cm²),

 $L = Length \ of \ sample \ (cm), \quad H = Difference \ of \ head \ (cm).$

Soil Porosity Determination: The soil was placed into a mould of volume $(V_T) = 1000 \text{ cm}^3$. The weights of the sample and mould (M_t) were determined. The sample is then dried in an oven at temperature 105°C and re-weights (M_s).

The weight of water M_w, weight of solid M_s, volume of voids V_v was determined based on equation 2 and 3 respectively.

 $G_s = specific \ gravity \ of \ solid, yw = density \ of \ water.$ Note G_s taken as 2.66 and $y = 1g/cm^3$.

RESULTS AND DISCUSSION

Borehole water turbidity (NTU) analysis results in Sagbama (SG) area is shown in Table 2. The turbidity averages over the sampled period is shown in Figure 1. Sagbama area borehole lithologic log is presented in Figure 2, while Sieve analysis and grain size distribution curve, (13m depth) of Sagbama area is presented in Figure 3.

Research Article

Table 2: Borehole Water Turbidity (NTU) Analysis Results in Sagbama (SG) Area

Day	SG1	SG2	SG3	SG4	SG5	SG6	SG7	SG8	SG9	SG10	SG11	Mean
0.00	2.55	4.46	2.53	3.41	5.01	4.00	4.00	4.30	4.00	4.00	3.13	3.76
1.00	2.34	4.54	2.98	3.21	4.86	4.00	4.00	4.40	4.00	4.00	2.56	3.72
15.00	2.34	4.34	2.70	3.44	4.50	4.00	4.00	4.01	4.00	4.00	3.34	3.70
29.00	2.44	3.54	3.00	3.45	4.34	4.00	4.00	3.56	4.00	4.00	3.45	3.62
43.00	2.23	4.43	3.10	2.54	4.33	4.00	4.00	3.34	4.00	4.00	3.23	3.56
57.00	3.10	4.54	3.12	3.34	4.12	4.00	2.00	3.23	5.00	6.00	3.32	3.80
71.00	2.23	4.43	3.02	3.45	4.01	4.00	2.00	3.16	4.00	4.00	2.35	3.33
85.00	2.31	4.43	2.98	3.32	4.02	5.00	4.00	3.23	5.00	2.00	3.22	3.59
99.00	1.98	3.43	2.79	3.31	4.10	2.00	5.00	3.10	2.00	2.00	2.33	2.91
113.00	2.73	3.43	3.10	3.32	4.33	2.00	5.00	3.41	4.00	4.00	3.32	3.51

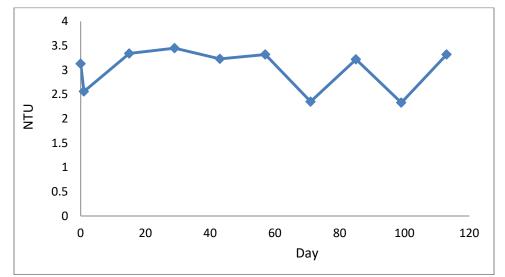


Figure 1: Turbidity Averages (NTU) over the Sampled Period in Sagbama Area

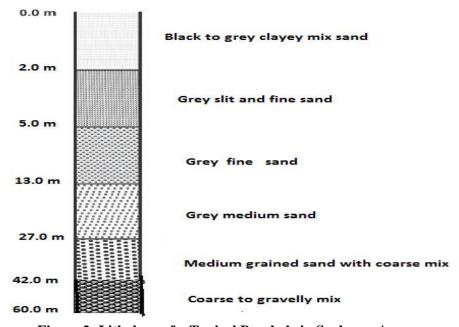


Figure 2: Lithology of a Typical Borehole in Sagbama Area

International Journal of Basic and Applied Chemical Sciences ISSN: 2277-2073 (Online) An Open Access, Online International Journal Available at http://www.cibtech.org/jcs.htm 2016 Vol. 6(4) October-December, pp.8-13/Madu et al.

Research Article

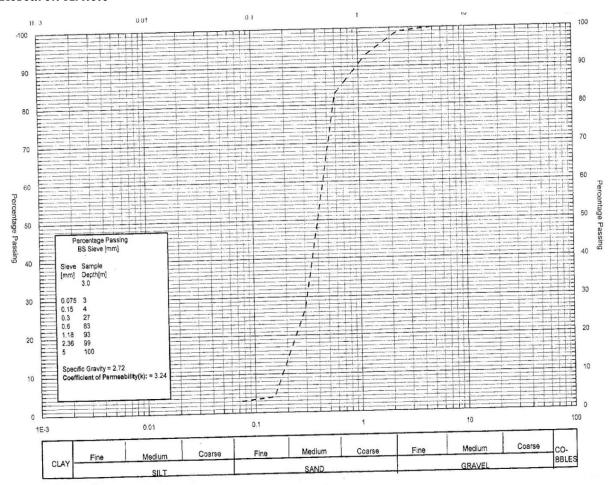


Figure 3: Sieve Analysis and Grain Size Distribution Curve, 3m Depth, Sagbama Area

The average turbidity value of the control (sample water before detonation) was 3.76 NTU. After dynamite detonation, the average measured turbidity values ranged from 2.91 to 3.80 NTU. The highest recorded average value of 3.80 NTU was obtained on day-57 after dynamite detonation. Thereafter the average turbidity value remained fluctuated. The lowest average value of 2.91 NTU was recorded on day-99. These variations from the control are not significant enough for dynamite to be said to have impacted on the turbidity value of the ground water. The control and test results values are both within the Federal Ministry of Environment, Housing and Urban Development (FMEnv&UD) limit of 10 NTU. Similar daily variations have been observed in areas where there was no dynamite detonation (Monteil-Rivera *et al.*, 2004).

The representative lithology of Sagbama area (Figure 2) as revealed by the borehole logging consists of 0-2m clayey sand, 2-5m silty and fine sand, 5-13m medium sand, 13-27m medium to coarse sand, 27 – 60m medium to coarse sands and gravelly mix (Figure 2). These litho-types are mainly non-plastics also categorized as cohensionless sands. The presence of silty sands at 4 to 5m depths could be an obstruction to infiltration of contaminants from dynamite detonation. Sieve analysis results from Sagbama (Figure 4) showed average to 7% passing for grain size 0.075mm, 8% passing for 0.15mm grain size, 39% passing for grain size 0.03mm, 77% passing for grain size 0.6 mm, 96% passing for grain size 2.36mm and 100% passing for grain size 5.00mm.

The Coefficient of permeability, K, at a depth of 3m (Figure 3) was 3.24 cm/sec. Permeability is the ability of a soil or rock type to conduct or discharge water (and effluents) under a hydraulic gradient. It depends on soil density, degree of saturation, viscosity and particle size. Infiltration capacity of soil

International Journal of Basic and Applied Chemical Sciences ISSN: 2277-2073 (Online) An Open Access, Online International Journal Available at http://www.cibtech.org/jcs.htm 2016 Vol. 6(4) October-December, pp.8-13/Madu et al.

Research Article

depends on permeability, degree of saturation, vegetation, amount of rainfall and duration of rainfall (Abam, 2004; Youdeowei, 2003). Coefficient of permeability of unconsolidated sand is affected by the fluid viscocity, grain size, grain sorting, grain shape and packing (Abam, 2004).

REFERENCES

Abam TKS (2004). *Geohyrology with Applications to Environmental Management.* (Charisma Graphics, Aba, Abia State, Nigeria) 147.

ATSDR (1995). *Toxicological Profile for 2,4,6-Trinitrotoluene*, Agency for Toxic Substances and Disease Registry, (U.S. Department of Health and Human Services, Public Health Service, Atlanta, Georgia).

Bausinger T and Preuss J (2005). Environmental remnants of the first World War: Soil contamination of a burning ground for arsenucal ammunition. *Bulletin of Environmental Contamination Toxicology* **74**(6) 1045-1052.

Bendinelli C (2009). Effect of Land mines and unexploded ordnance on the pediatric population and comparison with adults in rural Cambodia. *World Journal of Surgery* 33(5) 1070-1074.

Dimitrios K, Albert LJ, Raj B and Steve C (2011). Soils contaminated with explosives: Environmental fate and evaluation of state-of the-art remediation processes (IUPAC Technical Report). *Pure and Applied Chemistry* **83**(7) 1407–1484.

Dontsova KM, Pennington JC, Hayes C, Simunek J and Williforf CW (2009). Dissolution and transport of 2,4 DNT and 2,6 DNT from MI propellant in soil. *Chemosphere* **77**(4) 597-603.

Eisentraeger A, Reifferscheid G, Dardenne F, Blust R and Schofer A (2007). Hazard characterization and identification of a former ammunition site using microarrays, bioassays and chemical analysis. *Environmental Toxicology and Chemistry* **26**(4) 634-646.

Gray RH and McGrath DA (2007). Environmental monitoring at DOE's pantex plant in Amarillo, Texas. *Federal Facilities Environmental Journal* **6** 79-88

Juhasz AL and Naidu R (2007). Explosives: Fate, dynamics and ecological impact in terrestrial and marine environments. *Reviews of Environmental Contamamination and Toxicology* **191** 163-215.

Monteil-Rivera F, Beaulieu C, Deschamps S, Paguet L and Jalal H (2004). Determination of explosives in environmental water samples by solid- phase micro extraction-liquid chromatography. *Journal of Chromatography A* 1048 213-221.

Pennington JC and Brannon JM (2002). Environmental fate of explosives. *Thermochimica Acta* **384**(1-2) 163-172.

Youdeowei PO (2003). Evaluation of Environmental Pollution Susceptibility in Niger Delta Using Geotechnical Parameters. Ph.D Thesis, Rivers State University of Science and Technology, Portharcourt, Nigeria, 198.